

Production of mycosporine like amino acids (MAAs) by natural phytoplankton community in the Artic

Sun-Yong Ha

Korea Polar Research Institute (KOPRI), Division of Polar Ocean Sciences

Background



Ozone depletion in Arctic and Antarctica In 1985, Reported the 'Ozone hole' over \checkmark OMI Total Ozone for Aug 1, 2010 OMI Total Ozone for Nov 13, 2009 Antarctica. Ozone depletion still occurs and continue \checkmark 1. UV causes a chlorine atom to break way from CFC molecule the CFC molecule. NIVP_PMI_NASA_PM CSEC MIND-FMI-NAS COMI 125 Dobson Units Dark Gray < 100 and > 500 DU Stratosphere Sea ice vs. Open sea UV-B tranmission on the Sea ice $0.00 \quad 0.05 \quad 0.10 \quad 0.15 \quad 0.20 \quad 0.25$ 0.00 0.05 0.10 0.15 0.20 0.25 0.30 free chlorine W/m^2 W/m² CIO - chlorine free oxygen CIO - chloring O, - oxygen 0mfrom stratospher molecule 0m2. The free chlorine 3. The chlorine atom 4. A free oxygen 5. The result is atom hits an ozone pulls one oxygen atom atom hits the chlorine another free chlorine 0.15m molecule. monoxide molecule. away. atom. Free chlorine will continue to deplete ozone in the stratosphere. $\mathop{\rm Debth}_{0.6m}^{0.3m}$ 20m \checkmark In 2013, the average size of the ozone hole was 8.1 million square miles. 1m 40m 1.8m ✓ UV-B transmission on the different sea thickness 1m ice thickness and UVBR have 2m Open sea thickness 1.2m Sea ice thickness 1.8m transmitted deeper in open sea water. 60m



Background



Mycosporine-like amino acids (MAAs) as versatile elements Ecological **Function** Values Significance **Anti-Oxidant** Nitrogen storage Sunscreen Photosynthetic efficacy 310 370 400 220 250 280 340 protection Wavelength (nm) COOH 4-Deoxygadusol, Shinorine. CH₃ R = H (268 nm)R = H (334 nm)Porphyra-334, CH₂ $R = CH_3 (334 nm)$ ÓН **Proto-protection** HC OH **Thermal stress** но но COOH Mycosporine-Gly, R = COOH (310 nm) Usujirene, R = CH_3 (cis) CH₃ Mycosporine-taurine, R = $CH_2SO_3H(309 \text{ nm})$ CH₂ (357 nm) Palythene, Palythene, $R = CH_3$ (trans) HO HO **Osmotic pressure Desiccation stress** HO HC COOH NH

From: Gao and Garcia-Pichel 2011

Modified from Bhatia et al. 2011



Production rate of MAAs using HPLC





Collected each compounds



Measured the ¹³C value of each compounds

Separated MAA by HPLC

To calculated production rate of individual MAA

$$\Delta MAC(t) = MAC \times \frac{a_{\rm is} - a_{\rm ns}}{a_{\rm ic} - a_{\rm ns}}$$

 $\Delta MAC : The amount of each MAA carbon photosynthetically produced during the incubation$ $<math display="block">a_{is} : {}^{13}C atom \% in each MAA of incubated sample$ $<math display="block">a_{ns} : {}^{13}C atom \% in each MAA of natural sample$ $<math display="block">a_{ic} : {}^{13}C atom \% in {}^{13}C enriched inorganic carbon$ MAC : Concentration of each MAA carbon at the end of incubationBiolog

Ha et al. 2012 Photochemistry and Photobiology B: Biology 114: 1-14







- Production of MAAs of phytoplankton on the melting ponds, Arctic
- Distribution of MAAs along size-fractionated phytoplankton in the Beaufort Sea, Arctic
- Production rate of Phto-protective compounds on Labculture (*Porosira glacialis*)

Production of MAAs of phytoplankton on the melting ponds



Distribution of MAAs along size-fractionated phytoplankton in the Beaufort Sea, Arctic





- ✓ Micro-phytoplankton use the carbon to photoprotective mechanisms rather than cell reproduction.
- ✓ However, nano-, and pico-phytoplankton focus on to cell reproduction due to poor photoprotective mechanisms.

Production rate of Photo-protective pigment





- The time difference between the production of MAAs and the production of pigments is the direct result of the allocation of carbon to different photo-protective compounds
- We could find out the stable isotope probing method better than bulk concentration for interpreting the complex environment



- To reflect the synthetic pathways of photoprotective compounds and the carbon cycle within the cell in contrasting patterns over time that are defined by the production of photoprotective pigments and MAAs.
- The time difference between the production of MAAs and the production of pigments is the direct result of the allocation of carbon to different photoprotective compounds.
- Organic carbon is initially fixed to produce photoprotective pigments and that organic carbon (¹³C) is fixed to produce MAAs within the cell.

Thank you !!

√ This research was a part of the project titled 'K-AOOS (KOPRI, PM16040)', funded by the Ministry of Oceans and Fisheries, Korea.

Sun-Yong Ha : syha@kopri.re.kr

From Joo